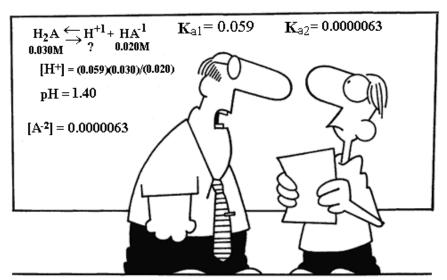
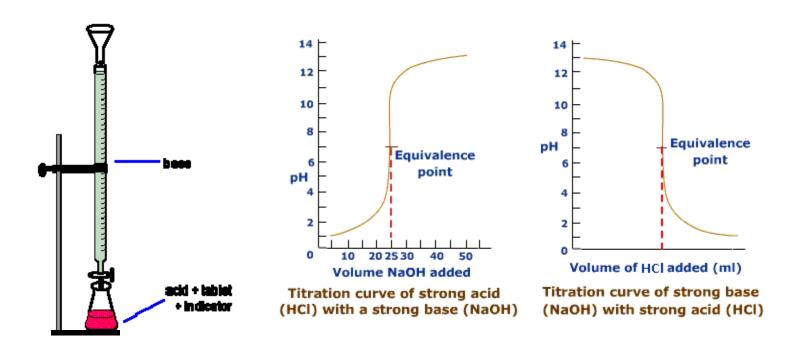


## **AP\* Chemistry Demystifying Titration Curves**

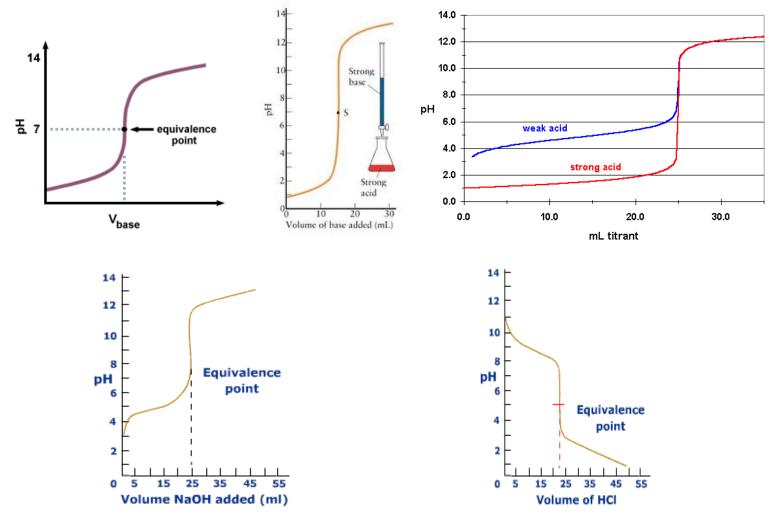
This handout is meant to accompany the instructional video found at either <a href="http://www.vimeo.com/19007364">http://www.vimeo.com/19007364</a> or <a href="http://www.edutube.org/en/video/ap-chemistry-interpreting-titration-curves">http://www.edutube.org/en/video/ap-chemistry-interpreting-titration-curves</a>. This won't make a lick of sense unless you watch the video containing the explanations!



Why? Because I had to learn this stuff and now it's your turn!

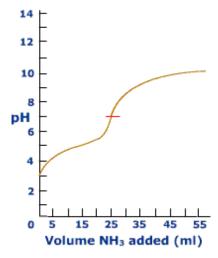


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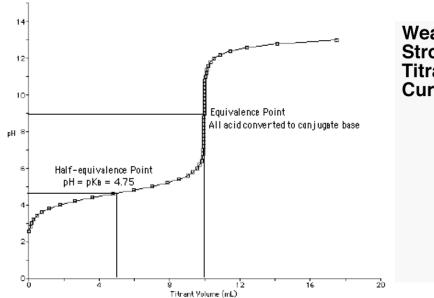
The pH titration curve of weak acid (CH<sub>3</sub>COOH) and strong base (NaOH)

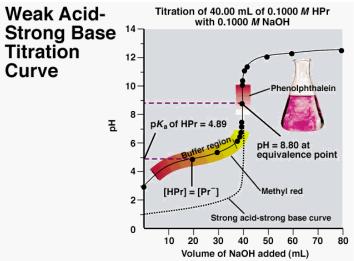
The pH titration curve of weak base(NH<sub>4</sub>OH) and strong acid (HCI)

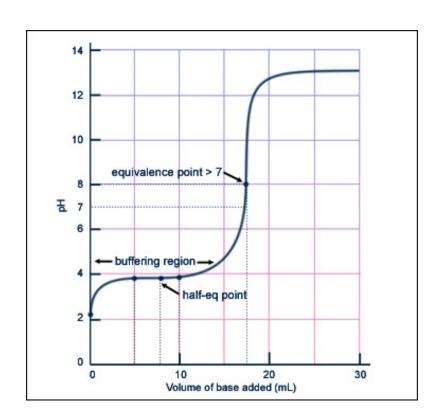


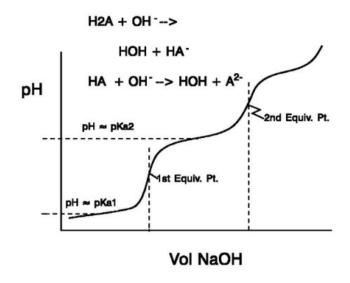
Titration curve of weak base (NH₄OH) and weak acid (CH₃COOH)

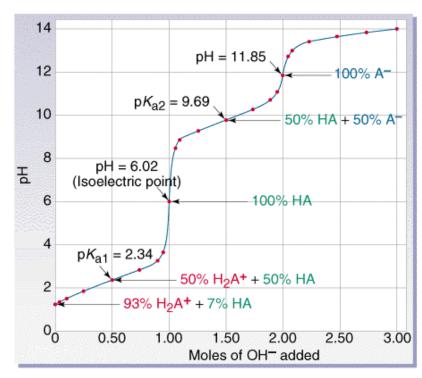
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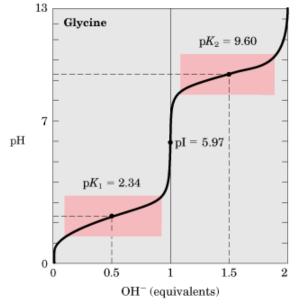


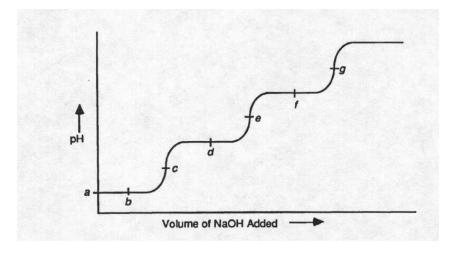












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